# **Electron Spin Polarizations of Phthalocyaninatosilicon Covalently Linked to One TEMPO Radical in the Excited Quartet and Doublet Ground States**

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Phthalocyaninatosilicon(IV) covalently linked to a 2,2,6,6-tetramethyl-1-piperidinyloxy (TEMPO) radical, SiPc-TEMPO, has been studied by time-resolved electron paramagnetic resonance (TREPR). A TREPR spectrum at 20 K is assigned to the doublet ground  $(D_0)$  state and the excited quartet  $(Q_1)$  state consisting of the excited triplet SiPc and the doublet TEMPO radical. The excited quartet spectrum is simulated using zero field splitting parameters  $D = 4.3 \times 10^{-3}$  and  $E = 3.3 \times 10^{-4}$  cm<sup>-1</sup>. The D value observed is almost identical with that calculated under a point charge approximation (=  $4.7 \times 10^{-3}$  cm<sup>-1</sup>). The electron spin polarization of the  $Q_1$  state is reproduced by selective intersystem crossing (ISC) from the excited doublet states to the  $\pm 1/2$  sublevels in the Q<sub>1</sub> state. This selective ISC is explained by spin-orbit coupling between the P<sub>x</sub> and P<sub>v</sub> orbitals on oxygen atoms of axial-ligands. In the TREPR spectrum at 293 K, SiPc-TEMPO molecules in the  $Q_1$  and  $D_0$  states exhibit A and E polarizations, respectively, where the A and E are absorption and emission of microwaves. The E polarization of the  $D_0$  state can be explained by the radical-quartet pair mechanism (RQPM), which originates from an intermolecular interaction between the  $Q_1$  and  $D_0$  SiPc-TEMPO molecules. This RQPM was observed for the first time. The intermolecular interaction between a stable radical and the Q<sub>1</sub> SiPc-TEMPO is clearly demonstrated using a toluene solution containing both SiPc-TEMPO and 2,6-ditert-butyl- $\alpha$ -(3,5-di-tert-butyl-4-oxo-2,5-cyclohexadien-1-ylidene)-p-tolyloxy (Galvinoxyl). The A and E polarizations are observed for the fast and slow components of Galvinoxyl, respectively, and can be assigned to the electron spin polarization transfer from the  $Q_1$  SiPc-TEMPO to  $D_0$  Galvinoxyl and the RQPM between the Q<sub>1</sub> SiPc-TEMPO and D<sub>0</sub> Galvinoxyl, respectively.

### Introduction

Using a time-resolved electron paramagnetic resonance (TREPR) technique, many kinds of intermediates generated by photoexcitation have been studied in terms of chemically induced dynamic electron spin polarizations (CIDEPs).<sup>1–7</sup> In these studies the electron spin polarizations (ESPs) of the excited triplet states,<sup>1–3</sup> spin-correlated radical pairs,<sup>4,5</sup> the triplet mechanism (TM),<sup>6</sup> and the radical pair mechanism (RPM)<sup>7</sup> have been intensively investigated and have provided important information for many photochemical reactions.

Recently new kinds of CIDEPs, the radical triplet pair mechanism (RTPM)<sup>8-10</sup> and the electron spin polarization transfer (ESPT) mechanism,11 have been reported, focusing on the interactions between photoexcited triplet species and stable radicals. To obtain novel information, some excited triplet molecules bonded to a stable radical were investigated.<sup>12,13</sup> Corvaja et al. studied C<sub>60</sub> covalently linked to a TEMPO radical  $(C_{60}$ -TEMPO) in toluene solution.<sup>12</sup> In this study, the transition between |+1/2> and |-1/2> in the lowest excited quartet (Q<sub>1</sub>) state, which was constituted by the excited triplet C<sub>60</sub> and doublet TEMPO, was observed for the first time. The CIDEP mechanism was explained by the mixing between the lowest excited doublet  $(D_1)$  and  $Q_1$  states originating from the magnetic dipole-dipole interaction. The CIDEP of C<sub>60</sub>-TEMPO in the doublet ground (D<sub>0</sub>) state was also observed, and originated from the ESP generated in the Q1 state. Namely, the CIDEPs of C60-TEMPO in the  $Q_1$  and  $D_0$  states were produced by the

and ESPT. Since  $Q_1$  molecules can have both ESPs and zero field splittings (zfs), the ESPT and radical-quartet pair mechanism (RQPM) will be produced by the interaction between the  $Q_1$  molecule and the stable radical in analogy with the radical-triplet system. However, there have been few studies focusing on the intermolecular interaction between the  $Q_1$ molecule and the stable radical. The one and only example is the metalloporphyrin coordinated by pyridyl nitronyl nitroxide radical (MP-nitpy) in toluene solution.<sup>13c</sup> Although the ESPT from the  $Q_1$  MP-nitpy to the nonligating nitpy was discussed in that study, there were several complexities to investigate the intermolecular interaction in solution: (1) Distinguishing between the nonligating nitpy and MP-nitpy was difficult. (2) Dissociation of the axial-ligating nitpy could occur easily in solution.

intramolecular interaction. In general, intermolecular interactions are important for CIDEPs in solution, i.e., the RPM, RTPM,

For these reasons, we have synthesized phthalocyaninatosilicon (SiPc) covalently linked to one 2,2,6,6-tetramethyl-1piperidinyloxy (TEMPO), SiPc-TEMPO (Figure 1), and studied its CIDEPs at 20 and 293 K by the TREPR method. The summary of important aspects in this research is as follows: (1) A TREPR spectrum of the  $Q_1$  state is observed at 20 K. The magnetic dipole–dipole interaction between the excited triplet SiPc and the doublet TEMPO is examined by comparison with calculation. The ESP mechanism is determined by a spectral simulation. This result is discussed in terms of spin– orbit coupling (SOC) between the  $Q_1$  and higher excited doublet states. (2) For toluene solutions of SiPc-TEMPO and those

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Figure 1. Molecular structures of  $SiPc(OH)_2$  (a), TEMPOL (b), Galvinoxyl (c), and SiPc-TEMPO (d).

containing both SiPc-TEMPO and 2,6-di-*tert*-butyl- $\alpha$ -(3,5-di*tert*-butyl-4-oxo-2,5-cyclohexadien-1-ylidene)-*p*-tolyloxy (Galvinoxyl), TREPR measurements are carried out to investigate the intermolecular interaction. For the SiPc-TEMPO system, the RQPM between the Q<sub>1</sub> and D<sub>0</sub> SiPc-TEMPO is observed for the first time. Further, the ESPT and RQPM between the Q<sub>1</sub> SiPc-TEMPO and D<sub>0</sub> Galvinoxyl are satisfactorily demonstrated for the SiPc-TEMPO–Galvinoxyl system.

#### **Experimental Section**

4-Hydroxy-TEMPO (TEMPOL) and Galvinoxyl were purchased from Aldrich Chemical Co. and Tokyo Chemical Industry Co., Ltd, respectively. (Dihydroxy)SiPc (SiPc(OH)<sub>2</sub>) and SiPc-TEMPO were synthesized following the methods already described.<sup>14</sup> Three signals split by the hyperfine coupling (hfc) of a nitrogen nucleus (I = 1) were observed in a steadystate EPR spectrum of SiPc-TEMPO at 293 K, and indicated the presence of a TEMPO radical.<sup>15</sup> Spectral grade toluene (Nacalai Tesque Inc.) was used as solvent, and the concentration of the samples was  $1 \times 10^{-3}$  M. All samples were deaerated by freeze-pump-thaw cycles.

TREPR measurements were carried out at 20 and 293 K on a Bruker ESP 300E spectrometer.<sup>13</sup> An Oxford ESR 900 cold gas flow system was used for the experiments at 20 K. Steadystate EPR measurements were made with the same apparatus. Samples were excited at 585 nm by a Lumonics HD 500 dye laser pumped with a Lumonics EX 500 excimer laser. The TREPR signals from the EPR unit were integrated by a LeCroy 9450A oscilloscope.

## **Results and Interpretations**

**TREPR Spectra in the Solid State.** TREPR spectra of SiPc-(OH)<sub>2</sub> and SiPc-TEMPO were observed at 20 K. A TREPR spectrum of SiPc(OH)<sub>2</sub> observed at 0.9  $\mu$ s after laser excitation is shown in Figure 2a (dotted line). From the spectral simulation (solid line), zfs parameters *D* and *E* of SiPc(OH)<sub>2</sub> in the lowest excited triplet (T<sub>1</sub>) state are evaluated as  $2.08 \times 10^{-2}$  and  $5.27 \times 10^{-3}$  cm<sup>-1</sup>, respectively.<sup>3</sup> The *A/E* polarization pattern indicates that intersystem crossing (ISC) from the S<sub>1</sub> state is selective to the T<sub>z</sub> sublevel (*z*: an out-of-plane axis), where the *A* and *E* denote absorption and emission of microwaves, respectively.

A TREPR spectrum of SiPc-TEMPO observed at 0.6  $\mu$ s after laser excitation is shown in Figure 2b (dotted line). The spectrum exhibits an *AAE* polarization pattern. The splitting between the



**Figure 2.** TREPR spectra (dotted lines) of SiPc(OH)<sub>2</sub> (a) and SiPc-TEMPO (b) with their simulations (solid lines) and a steady-state EPR spectrum of SiPc-TEMPO (c). TREPR spectra were observed at 0.9 (a) and 0.6  $\mu$ s (b) after laser excitation. These spectra were observed at 20 K.

outer *A* and *E* signals of SiPc-TEMPO is much smaller than that of SiPc(OH)<sub>2</sub>. The outer *A/E* signals of SiPc-TEMPO are similar to those of ZnTPP-nitpy and are assigned to the excited quartet state. For a quantitative discussion, a spectral simulation was carried out. The method of simulation of excited quartet spectra has already been reported.<sup>13b</sup> A simulation spectrum is shown in Figure 2b (solid line) and reproduces the observed *A/E* signals. The *D* and *E* values of the quartet state are estimated as  $4.3 \times 10^{-3}$  and  $3.3 \times 10^{-4}$  cm<sup>-1</sup>, respectively. The simulation suggests that the selectivity in the ISC is P(+3/2):P(+1/2): P(-1/2):P(-3/2) = 0:1:1:0. To confirm the validity of the zfs parameter, the *D* value was calculated using the following equation.<sup>13,16</sup>

$$D(Q_1) = D(T_1)/3 + D(RT_1)/3$$
 (1)

 $D(T_1)$  was evaluated from the TREPR spectrum of SiPc(OH)<sub>2</sub>.  $D(RT_1)$  was calculated under a point charge approximation.<sup>17</sup> By diagonalizing  $D(Q_1)$ , the  $D(Q_1)$  value is calculated as 4.7 ×  $10^{-3}$  cm<sup>-1</sup>. The *D* value calculated is very close to that observed (= 4.3 ×  $10^{-3}$  cm<sup>-1</sup>). From this calculation, the outer *A* and *E* signals can be assigned to the Q<sub>1</sub> state consisting of SiPc in the T<sub>1</sub> state (<sup>3</sup>SiPc<sup>\*</sup>) and TEMPO in the doublet ground (D<sub>0</sub>) state (<sup>2</sup>TEMPO).

In contrast the central A signal ( $g = 2.007 \pm 0.002$ ) is assigned to the doublet state, since it cannot be reproduced by the excited quartet state.<sup>13b</sup> To assign the doublet state, the *g* value of the excited doublet state (*g*(D<sub>1</sub>)), which consists of <sup>3</sup>SiPc\* and <sup>2</sup>TEMPO, was calculated as follows.<sup>13,16</sup>

$$g(D_1) = -1/3g(R) + 4/3g(T_1)$$
(2)

 $g(T_1)$  was evaluated as 2.000  $\pm$  0.003 from the experiment and g(R) was evaluated as 2.006 from the D<sub>0</sub> TEMPOL. As a result,



**Figure 3.** A steady-state EPR (a) and TREPR spectra (b-d) of SiPc-TEMPO. TREPR spectra were observed at 0.1 (b), 1.9 (c), and 3.6  $\mu$ s (d) after laser excitation. These spectra were observed at 293 K.

 $g(D_1)$  is calculated as 1.998 ± 0.003, and is different from the *g* value observed. A steady-state EPR spectrum of SiPc-TEMPO at 20 K is shown in Figure 2c. It is found that the *g* value (= 2.007 ± 0.002) of the *A* signal is almost identical with that (= 2.005 ± 0.001) of the maximal peak position in the steady-state EPR spectrum. Therefore, this *A* polarization is assigned to the D<sub>0</sub> state.

**TREPR Spectra in Solution.** TREPR signals of SiPc-TEMPO were observed at 293 K, and TREPR spectra at 0.1, 1.9, and 3.6  $\mu$ s are shown in Figure 3 with a steady-state EPR spectrum. A spectral simulation was carried out typically for the TREPR spectrum at 1.9  $\mu$ s. The spectrum is reproduced by using both a broad A signal ( $g = 2.004 \pm 0.002$ ) and three sharp E signals ( $g = 2.0060 \pm 0.0002$ , hfcc = 15.4 G), as shown in Figure 3c (solid line). The three sharp E signals are assigned to the D<sub>0</sub> state from the g and hfcc values, while the broad A signal is obviously not. The observable excited states in our TREPR experiment (time resolution of 100 ns) are the D<sub>1</sub> and Q<sub>1</sub> states formed by <sup>3</sup>SiPc\* and <sup>2</sup>TEMPO. Since the calculated  $g(D_1)$  value (= 1.998) is different from the observed g value, the possibility of the D<sub>1</sub> state can be ruled out. The g value of the Q<sub>1</sub> state ( $g(Q_1)$ ) is expressed as follows.<sup>13,16</sup>

$$g(Q_1) = 1/3g(R) + 2/3g(T_1)$$
 (3)

Using eq 3,  $g(Q_1)$  is calculated to be  $2.002 \pm 0.003$ , which is close to the observed g value (=  $2.004 \pm 0.002$ ). Therefore, this broad A signal can be assigned to the  $Q_1$  state. This assignment is supported by the fact that the  $Q_1$  state in solution is observed for several systems,<sup>12,13c</sup> while the D<sub>1</sub> state has not been observed in solution because of the short lifetime.



**Figure 4.** A steady-state EPR spectrum (a) and TREPR spectra (b, c) for the SiPc-TEMPO–Galvinoxyl system. TREPR spectra were observed at 0.6 (b) and 3.4  $\mu$ s (c) after laser excitation. These spectra were observed at 293 K.

TREPR measurements were carried out for toluene solutions containing both SiPc-TEMPO and Galvinoxyl radical in order to evaluate interactions between the Q1 SiPc-TEMPO and a stable radical. A steady-state EPR spectrum of the SiPc-TEMPO–Galvinoxyl system is shown in Figure 4a. Since the EPR spectrum of Galvinoxyl is obviously different from that of SiPc-TEMPO, the SiPc-TEMPO-Galvinoxyl system is appropriate for this study. TREPR spectra of the SiPc-TEMPO-Galvinoxyl system at 0.6 and 3.4  $\mu$ s are shown in Figures 4b and 4c, respectively. Two kinds of signals, a broad A signal and several sharp A signals, are observed in the TREPR spectrum at 0.6  $\mu$ s.<sup>18</sup> By comparison with the SiPc-TEMPO system, the broad A signal is assigned to the Q<sub>1</sub> SiPc-TEMPO. Since the resonance magnetic fields of the several sharp Asignals are consistent with those of the steady-state EPR spectrum of Galvinoxyl, the sharp TREPR signals are assigned to the  $D_0$  Galvinoxyl with ESPs. On the other hand, several E signals are observed in the TREPR spectrum at 3.4  $\mu$ s. By comparison with the steady-state EPR spectrum, these sharp signals are assigned to the D<sub>0</sub> SiPc-TEMPO and D<sub>0</sub> Galvinoxyl.

#### Discussion

**CIDEP in the Solid State.** The spectral simulation of SiPc-TEMPO has revealed that the ISC rates from the higher excited doublet states to the  $|\pm 1/2\rangle$  sublevels in the Q<sub>1</sub> state are faster than those to the  $|\pm 3/2\rangle$  sublevels. This selectivity originates from coupling between the spin and orbital angular momentum along the out-of-plane axis.<sup>13</sup> To confirm this origin, the SOC between the excited doublet states and the Q<sub>1</sub> state is examined. Since the excitation energy of TEMPO (~20000 cm<sup>-1</sup>) is much higher than those of the S<sub>1</sub> (~14500 cm<sup>-1</sup>) and T<sub>1</sub> (~9000 cm<sup>-1</sup>) SiPc, the excited states constituted by the S<sub>1</sub> or T<sub>1</sub> SiPc have to be considered. In the ZnTPP-nitpy system, the ISC rates to the  $|\pm 1/2\rangle$  in the Q<sub>1</sub> state are faster than those to the  $|\pm 3/2\rangle$ . This is explained by the SOC due to the  $d_{\pi}$  orbitals on the zinc ion.<sup>13</sup> However, the  $d_{\pi}$  orbitals cannot influence the properties of the  $S_1$  and  $T_1$  SiPc, since the energies of  $d_{\pi}$  orbitals on the Si ion are much higher than that of the LUMO of phthalocyanines. Alternatively, SOC on the oxygen atoms of the axial-ligands is possible. Gouterman et al. studied the excited triplet states of porphyrins and noted the importance of the SOC between the  $P_x$  and  $P_y$  orbitals of the axial-ligands.<sup>19</sup> The fact that the SOC between the  $P_x$  and  $P_y$  orbitals on the oxygen atoms is more efficient than that along the in-plane axes is also manifested in the selective ISC of  $SiPc(OH)_2$  in this study. Hence, the ESP of the Q<sub>1</sub> SiPc-TEMPO can be reasonably interpreted by the SOC between the  $P_x$  and  $P_y$  orbitals on the oxygen atoms of the axial-ligands.

The A polarization of the  $D_0$  state was observed in the solid state and revealed that the  $\beta$  spin was generated selectively in the  $Q_1$  state. It has been reported that for the  $C_{60}$ -TEMPO system the ESP in the  $Q_1$  state is transferred to the  $D_0$  state via an intramolecular process.<sup>12</sup> Therefore, the A polarization of the  $D_0$  state is considered to be due to the excess  $\beta$  spin generated in the  $Q_1$  state.

**CIDEP in Solution.** For the SiPc-TEMPO system at 293 K, the *A* and *E* polarizations were observed for the  $Q_1$  and  $D_0$  SiPc-TEMPO, respectively. The *A* polarization of the  $Q_1$  state in solution is consistent with the ESP in the solid state, where the  $\beta$  spin is generated selectively. This *A* polarization is explained by selective ISC. On the other hand, the *E* polarization of the  $D_0$  state observed in solution is contrary to the *A* polarization of the  $D_0$  state observed in the solid state. The origins of this *E* polarization must be explained.

When SiPc-TEMPO molecules are partially excited, three electronic states, the D<sub>0</sub>, D<sub>1</sub>, and Q<sub>1</sub> states, are formed in solution. For this phenomenon, two kinds of CIDEP mechanisms due to intramolecular interactions are conceivable. One is the CIDEP of the Q<sub>1</sub> state produced by SOC. Although excess  $\beta$  spin is generated in the Q<sub>1</sub> state as seen in the experiments at 20 and 293 K, this  $\beta$  spin does not provide the *E* polarization of the D<sub>0</sub> state at 293 K. The second is the mixing between the Q<sub>1</sub> and D<sub>1</sub> states by magnetic dipole–dipole interaction. This polarization mechanism was observed for some C<sub>60</sub>-TEMPO systems,<sup>12</sup> where the ESP of the Q<sub>1</sub> state was transferred to the D<sub>0</sub> state and was the same as that of the D<sub>0</sub> state is the opposite polarization of the Q<sub>1</sub> state. Therefore, it cannot be interpreted by the intramolecular interaction.

Next intermolecular interactions need to be considered. For the interaction between the  $D_1$  and  $D_0$  states, the RPM is possible. However, since the hyperfine dependence is insignificant, the RPM can be neglected.<sup>7</sup> This is supported by the fact that the rate of polarization of the D<sub>0</sub> state is slower than that of the Q1 state, despite the shorter lifetime of the D1 state. For the interaction between the  $Q_1$  and  $D_0$  states, two kinds of CIDEPs are probable. One of which is the ESPT from a  $\ensuremath{Q_1}$ molecule to a  $D_0$  molecule.<sup>11,13c</sup> The ESP of the  $Q_1$  state is the A polarization, while that of the  $D_0$  state is the *E* polarization. The E polarization of the D<sub>0</sub> SiPc-TEMPO, therefore, cannot be interpreted by the ESPT. The other is the RQPM, which is similar to the RTPM.8 When a D<sub>0</sub> molecule interacts with a Q<sub>1</sub> molecule, the triplet (T) and quintet (F) states are generated as shown in Figure 5. When J < 0 (J: an electron exchange interaction), an E polarization is expected by zfs interactions in analogy with the RTPM. The insignificant hyperfine depen-



Figure 5. The radical quartet pair mechanism.

dence observed is similar to that in the RTPM. Consequently, the E polarization of the D<sub>0</sub> state in solution can be assigned to the RQPM.

For the SiPc-TEMPO-Galvinoxyl system, the intermolecular interaction between  $Q_1$  and  $D_0$  molecules is nicely substantiated. The fast and slow components of the CIDEPs of Galvinoxyl were the A and E polarizations, respectively. These ESPs are generated by the interaction between the Q<sub>1</sub> SiPc-TEMPO and the D<sub>0</sub> Galvinoxyl and are interpreted with reference to the case of the radical-triplet system. When a stable radical interacts with an excited triplet molecule, two kinds of CIDEPs, i.e., the ESPT and RTPM, are generated. The ESPT occurs initially, and the ESP due to the RTPM is generated after spin lattice relaxation.<sup>11</sup> For the radical-quartet system the ESPT and RQPM are considered. Since the ESP of the Q1 SiPc-TEMPO is the A polarization, an A polarization of radicals will be generated by the ESPT mechanism. In contrast, when J < 0, the RQPM will provide an *E* polarization. Accordingly, the *A* and *E* polarizations will be produced in that order. Indeed, the fast A and slow Epolarizations of the D<sub>0</sub> Galvinoxyl were observed in the SiPc-TEMPO-Galvinoxyl system. As a result, these A and E polarizations are assigned to the ESPT and RQPM generated by the intermolecular interaction between the Q1 SiPc-TEMPO and D<sub>0</sub> Galvinoxyl. The CIDEP mechanism of the SiPc-TEMPO–Galvinoxyl system is summarized in Figure 6.<sup>20</sup> The A polarization of the Q<sub>1</sub> SiPc-TEMPO is generated by selective ISC from the higher excited doublet states  $(D_n)$ . The ESPT occurs from the  $Q_1$  SiPc-TEMPO with the A polarization to the  $D_0$  Galvinoxyl. After spin lattice relaxation, the *E* polarizations of the D<sub>0</sub> SiPc-TEMPO and Galvinoxyl are produced by the RQPM.

## Conclusions

The CIDEPs of SiPc-TEMPO in solution and in the solid state were studied by TREPR. The TREPR spectrum of SiPc-TEMPO in the solid state was assigned to the D<sub>0</sub> state and the Q<sub>1</sub> state, which consists of <sup>3</sup>SiPc\* and <sup>2</sup>TEMPO. The *D* value of the Q<sub>1</sub> state evaluated by the spectral simulation was almost the same as that calculated under the point charge approximation. The ESP in the Q<sub>1</sub> state was reproduced by selective ISC from the excited doublet states to the  $|\pm 1/2>$  sublevels in the Q<sub>1</sub> state. This ISC was explained by the SOC between the P<sub>x</sub> and P<sub>y</sub> orbitals on the oxygen atoms of the axial-ligands.

For the SiPc-TEMPO system, two components, a broad *A* signal and three sharp *E* signals, were observed at 293 K and were assigned to the  $Q_1$  and  $D_0$  states, respectively. The *E* 



Intramolecular Process Intermolecular Process

Figure 6. Summary of the CIDEP mechanism for the SiPc-TEMPO-Galvinoxyl system.

polarization of the  $D_0$  state could be interpreted in terms of the radical-quartet pair mechanism. Further, the ESPT and RQPM generated by the intermolecular interaction between the  $Q_1$  SiPc-TEMPO and  $D_0$  Galvinoxyl were nicely substantiated for the SiPc-TEMPO–Galvinoxyl system.

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#### **References and Notes**

(1) (a) Kim, S. S.; Weissman, S. I. J. Magn. Reson. **1976**, 24, 167. (b) Kim, S. S.; Weissman, S. I. Rev. Chem. Intermed. **1979**, 3, 107.

(2) (a) Murai, H.; Imamura, T.; Obi, K. *Chem. Phys. Lett.* **1982**, *87*, 295. (b) Murai, H.; Imamura, T.; Obi, K. *J. Phys. Chem.* **1982**, *86*, 3279.
(c) Terazima, M.; Yamauchi, S.; Hirota, N. *Chem. Phys. Lett.* **1983**, *98*, 145.

(3) (a) Chandrashekar, T. K.; van Willigen, H.; Ebersole, M. H. J. Phys. Chem. 1984, 88, 4326. (b) Gonen, O.; Levanon, H. J. Phys. Chem. 1985, 89, 1637. (c) Ishii, K.; Yamauchi, S.; Ohba, Y.; Iwaizumi, M.; Uchiyama, I.; Hirota, N.; Maruyama, K.; Osuka, A. J. Phys. Chem. 1994, 98, 9431. (d) Ishii, K.; Ohba, Y.; Iwaizumi, M.; Yamauchi, S. J. Phys. Chem. 1996, 100, 3839.

(4) (a) Sakaguchi, Y.; Hayashi, H.; Murai, H.; I'Haya, Y. J. Chem. Phys. Lett. **1984**, 110, 275. (b) Sakaguchi, Y.; Hayashi, H.; Murai, H.; I'Haya, Y. J.; Mochida, K. Chem. Phys. Lett. **1985**, 120, 401. (5) (a) Buckley, C. D.; Hunter, D. A.; Hore, P. J.; McLauchlan, K. A. *Chem. Phys. Lett.* **1987**, *135*, 307. (b) Tominaga, K.; Yamauchi, S.; Hirota, N. J. Chem. Phys. **1990**, *92*, 5175.

(6) (a) Wong, S. K.; Hutchinson, D. A.; Wan, J. K. S. J. Chem. Phys. **1973**, 58, 985. (b) Pedersen, J. B.; Freed, J. H. J. Chem. Phys. **1975**, 62, 1706.

(7) (a) Adrian, F. J. J. Chem. Phys. **1971**, 54, 3918. (b) Adrian, F. J.; Monchick, L. J. Chem. Phys. **1979**, 71, 2600.

(8) (a) Blättler, C.; Jent, F.; Paul, H. Chem. Phys. Lett. 1990, 166, 375. (b) Kawai, A.; Okutsu, T.; Obi, K. J. Phys. Chem. 1991, 95, 9130. (c) Kawai, A.; Obi, K. J. Phys. Chem. 1992, 96, 52. (d) Kawai, A.; Obi, K. Res. Chem. Intermed. 1993, 19, 865. (e) Turro, N. J.; Khudyakov, I. V.; Bossmann, S. H.; Dwyer, D. W. J. Phys. Chem. 1993, 97, 1138.

(9) (a) Corvaja, C.; Franco, L.; Pasimeni, L.; Toffoletti, A.; Montanari,
L. Chem. Phys. Lett. 1993, 210, 355. (b) Corvaja, C.; Franco, L.; Toffoletti,
A. Appl. Magn. Reson. 1994, 7, 257. (c) Corvaja, C.; Franco, L.; Pasimeni,
L.; Toffoletti, A. J. Chem. Soc., Faraday Trans. 1994, 90, 3267.

(10) (a) Hugerat, M.; van der Est, A.; Ojadi, E.; Biczok, L.; Linschitz, H.; Levanon, H.; Stehlik, D. J. Phys. Chem. **1996**, 100, 495. (a) Regev. A.; Galili, T.; Levanon, H. J. Phys. Chem. **1996**, 100, 18502.

(11) (a) Fujisawa, J.; Ishii, K.; Ohba, Y.; Iwaizumi, M.; Yamauchi, S. *J. Phys. Chem.* **1995**, *99*, 17082. (b) Fujisawa, J.; Ohba, Y.; Yamauchi, S. *J. Phys. Chem. A* **1997**, *101*, 434. (c) Jenks, W. S.; Turro, N. J. *Res. Chem. Intermed.* **1990**, *13*, 237.

(12) (a) Corvaja, C.; Maggini, M.; Prato, M.; Scorrano, G.; Venzin, M. *J. Am. Chem. Soc.* **1995**, *117*, 8857. (b) Corvaja, C.; Maggini, M.; Ruzzi,
M.; Scorrano, G.; Toffoletti, A. *Appl. Magn. Reson.* **1997**, *12*, 477.

(13) (a) Ishii, K.; Fujisawa, J.; Ohba, Y.; Yamauchi, S. J. Am. Chem. Soc. 1996, 118, 13079. (b) Ishii, K.; Fujisawa, J.; Adachi, A.; Yamauchi, S.; Kobayashi, N. J. Am. Chem. Soc. 1998, 120, 3152. (c) Fujisawa, J.; Ishii, K.; Ohba, Y.; Yamauchi, S.; Fuhs, M.; Möbius, K. J. Phys. Chem. A 1997, 101, 5869.

(14) (a) Ishii, K.; Hirose, Y.; Kobayashi, N. J. Am. Chem. Soc. **1998**, 120, 10551. (b) Ishii, K.; Hirose, Y.; Kobayashi, N. J. Porphyrins Phthalocyanines, in press.

(15) The g value and hfc constant (hfcc) of SiPc-TEMPO were evaluated as  $2.0060 \pm 0.0002$  and 15.4 G, and were the same as those of TEMPOL. The EPR signal of SiPc-TEMPO at higher field was weaker than that at lower field. This is reasonably explained by slower molecular rotation due to the attached SiPc (Weil, J. A.; Bolton, J. R.; Wertz, J. E. *Electron Paramagnetic Resonance Elementary Theory and Practical Applications*; Wiley-Interscience: New York, 1994).

(16) Bencini, A.; Gatteschi, D. EPR of Exchange Coupled Systems; Springer-Verlag: Berlin, 1990.

(17) (a) Stewart, J. J. P. J. Comput. Chem. 1989, 10, 209. (b) Mooney,
 J. R.; Choy, C. K.; Knox, K.; Kenney, M. E. J. Am. Chem. Soc. 1975, 97, 3033.

(18) A TREPR spectrum of Galvinoxyl observed at 0.2  $\mu$ s after laser excitation was slightly different from that at 0.6  $\mu$ s. This is considered to originate from a complex consisting of SiPc-TEMPO and Galvinoxyl in the excited state.

(19) Gouterman, M. In *The Porphyrins*; Dolphin, D., Ed.; Academic Press: New York, 1978; Vol. III, pp 1–165.

(20) Two reasons are considered for the undetectable A polarization of the  $D_0$  SiPc-TEMPO due to the ESPT. One is concentration. Since SiPc-TEMPO molecules are partially excited, the concentration of the  $D_0$  SiPc-TEMPO is low. The other is molecular size. For a bulky SiPc moiety, the  $D_0$  SiPc-TEMPO cannot be in contact with the  $Q_1$  SiPc-TEMPO contrary to Galvinoxyl. This consideration is supported by a previous report, where an importance of close contact is discussed for the ESPT between the excited triplet porphyrins and doublet radicals.<sup>11b</sup>